

# The Ring Launching Solar Cell: The RING matters

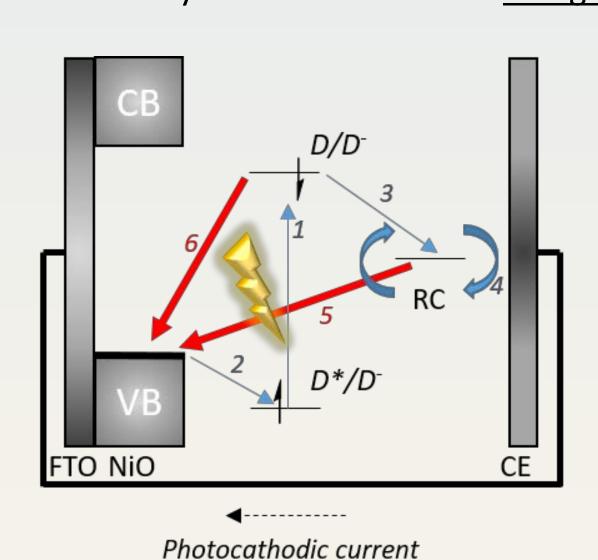


## Tessel Bouwens, Jenny Hasenack, Simon Mathew, Joost N. H. Reek

Supramolecular & Homogeneous Catalysis Group, van 't Hoff Institute for Molecular Sciences, University of Amsterdam, Science Park 904, 1098 XH Amsterdam

## Introduction

Today, n-type Dye-sensitized solar cells (DSSCs) are a well-established exceeding 14% photonconversion efficiency (PCE). The complementary p-type DSSC opens new avenues to tandem solar cells to have greater PCE for future application in both photovoltaics and solar-driven fuel generation. However, p-type DSSCs are relatively unexplored and the efficiency remains low due to charge recombination.



Blue arrows => the forward electron propagation red arrows => the unfavorable recombnation

How to solve this recombination issue?

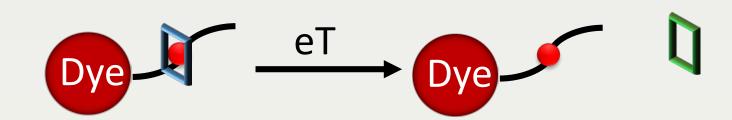
Spatial separation of charges

 $\rightarrow$  Donor- $\pi$ -Acceptor

Alternative strategy:

Vectorial electron transfer

→ Supramolecular approach of pre-organization of the redox mediator



High affinity when oxidized

Low affinity upon reduction<sup>2</sup>

Pre-organization of the redox mediator leads to increased photocurrents in DSSCs

What is the difference between a cationic or neutral ring in photovoltaic properties of the solar cell?

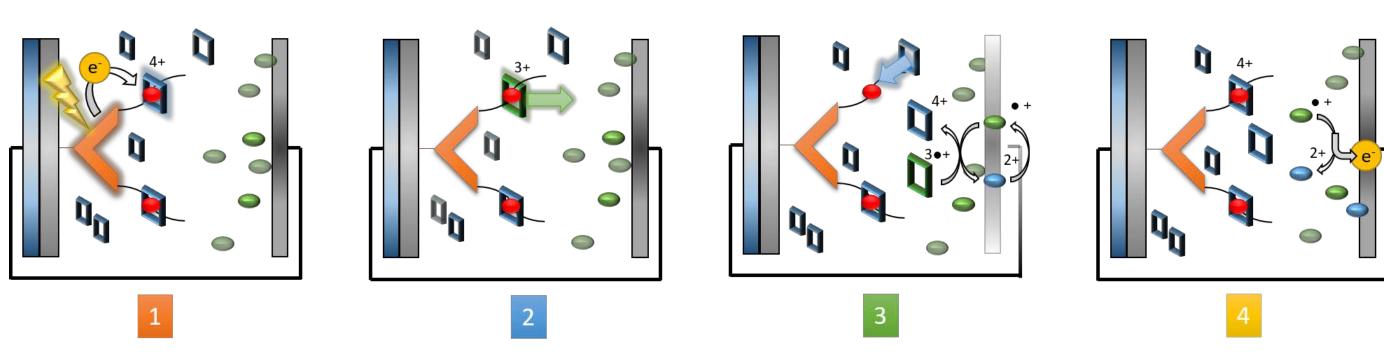
#### Design 2. Neutral Ring 1. Cationic Ring • NDI–RING has a low binding affinity $\rightarrow$ (~100 M<sup>-1</sup>) The reduction potential is more negative than MV-RING<sup>4+</sup> > (380 mV and 50 mV versus NHE) Binding studies show that • The RING<sup>4+</sup> has a high binding affinity for the DNP recognition sites: $K_{ass} = 3.4 \times 10^4 \, M^{-1}$ $\alpha = 0.7$ • The RING<sup>4+</sup> is able to quench in total 83% of the toal fluorescence of the dye at 630 nm. FTO NiO Photovoltaic properties Photovoltaic properties Dye 100 mW ⋅ cm<sup>-2</sup> irridiation 50 mW ⋅ cm<sup>-2</sup> irridiation • V<sub>OC</sub> = 307 • FF = 0.35 • $\eta = 0.02\%$ ---- P<sub>N</sub> (0.3% RING<sup>4+/3+</sup> + MV<sup>2+/+</sup> 0.00 0.05 0.10 0.15 0.20 0.25 0.30 MV–RING<sup>4+</sup> > Pre-organization of the redox mediator by

Wavelength (nm)

## Proposed operational mode

pseudorotaxane formation leads to DSSC exhibiting

10 x higher photocurrents.



Excitation of the  $P_N$  dye leads to eT to the RING4+.

Reduction of **RING**<sup>4+</sup> to RING<sup>3+</sup> leads to ring dissociation.

Upon dissociation RING4+

will bind the the DNP and

the e<sup>-</sup> can hop from

 $RING^{3+}$  to  $MV^{2+}$ .

Regeneration of the redox couple to complete the circuit.

The operational principle of the **NDI-RING** is supposed to proceed via a similar mechanism

## Conclusions and Outlook

- New type of p-type DSSC is made where the redox acceptor is preorganized to the dye. This  $P_N$  dye is based on the P1 dye functionalized with DNP moeities that can form pseudorotaxanes with the electron acceptor **MV-RING**<sup>4+</sup> ( $K_{ass} = 3 \times 10^4 \text{ M}^{-1}$ ) or **NDI-RING** ( $K_{ass} \approx 1 \times 10^2 \text{ M}^{-1}$ ).
- Solar cells based on pseudorotaxanes afforded photocurrents of one order of magnitude higher compared to the control experiments (no pseudorotaxane formation).
- Neutral **NDI–RING** shows 50 x higher photocurrents than the cationic **MV–** RING<sup>4+</sup>.
- Future research will focus on characterization and optimization of system



- NDI-RING

<sup>1.</sup> B. O'Regan, M. Gratzel, Nature, 1991, 353, 737-740.

<sup>2.</sup> P. R. Ashton, V. Balzani, O. Kocian, L. Prodi, N. Spencer and J. F. Stoddart, J. Am. Chem. Soc., 1998, 120, 11190–11191.

<sup>3.</sup> P. Qin, H. J. Zhu, T. Edvinsson, G. Boschloo, A. Hagfeldt and L. C. Sun, J. Am. Chem. Soc., 2008, 130, 17629. 4. T. Bouwens, S. Mathew, J. N. H. Reek, Faraday Discuss., 2019, accepted.